Date:

- 1. Identify each change below as a physical, chemical, or nuclear change. For each, justify your answer.
- a) During intermission at a hockey game, a zamboni sprays warm water onto the ice surface. This water freezes to form a slick new surface.

Type of change:



How do you know what type of change it is? ____

b) Deuterium Helium

Type of change:

How do you know what type of change it is?

c) A Coleman stove heats up as the propane burns.

$$C_3H_8 + O_2 \longrightarrow CO_2 + H_2O$$

Type of change:



How do you know what type of change it is?

2	Given	the	following	equation	for a	nuclear	disintegra	ition:
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?
$$\longrightarrow_{81}^{207} TI + {}_{2}^{4} He$$

a) What type of radiation is emitted during this decay process?

b) Name the unknown element. Explain how you got your answer.

Unknown element: _____

Explanation:

c) What is the mass number of this element? Explain your answer, based on the principle of conservation of mass.

Mass number: _____

Explanation:

3. The equation for the nuclear disintegration of S-35 is the following:

a) What type of radiation is emitted during this decay process?

b) Using the principle of conservation of mass, explain why the mass number remains unchanged.

c) Explain why the atomic number increases from 16 to 17.
