PSC-4012 Quiz # 9

Name:

Date:

1. How many moles of oxygen are required to react with 30 grams of glucose, $C_6H_{12}O_6$?

$$C_6H_{12}O_6 + O_2 \rightarrow CO_2 + H_2O$$

2. If 15 moles of fluorine gas react then what mass of dinitrogen tetrafluoride will be produced?

$$F_2 + NH_3 \longrightarrow N_2F_4 + HF$$

Bonus: How many grams of water are produced if 3mol of ammonia (nitrogen trihydride) are used?

$$NH_3 + NO \rightarrow N_2 + H_2O$$