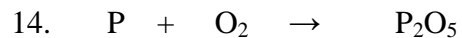
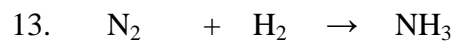
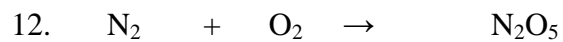
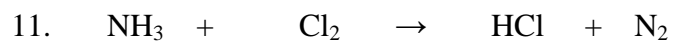
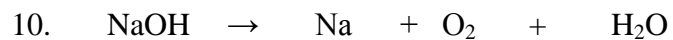
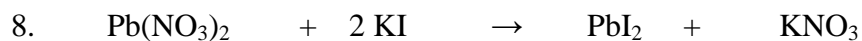
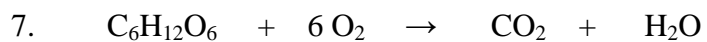
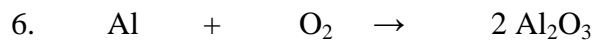
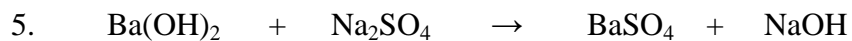
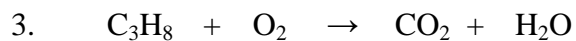
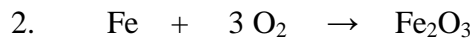
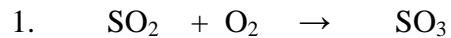
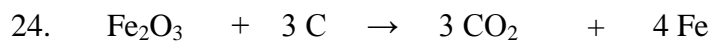
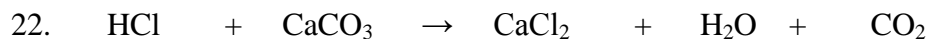
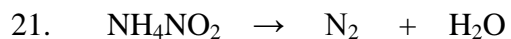
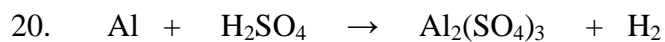
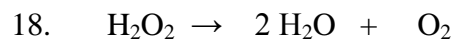
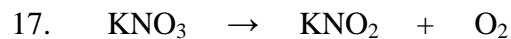
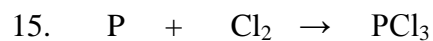


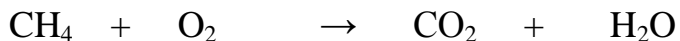
Balancing Equations Practice #1



25. The neutralization of hydrochloric acid (HCl) by calcium carbonate (CaCO_3) produces calcium chloride (CaCl_2), carbon dioxide (CO_2) and water (H_2O).

Write the balanced equation for this neutralization reaction.

26. The combustion of methane, CH_4 mixed with O_2 produces carbon dioxide, CO_2 , and water, H_2O . The unbalanced chemical equation for this reaction is as follows:



Balance the chemical equation for this combustion reaction.