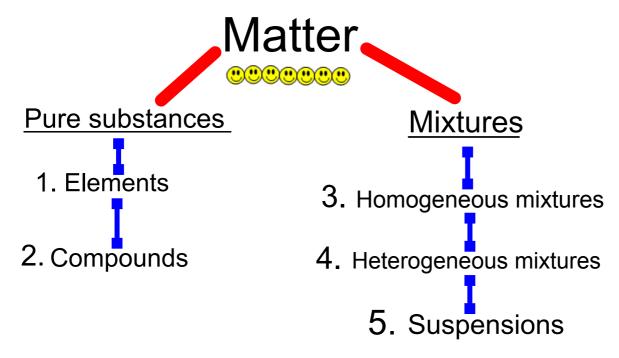
Today we will be learning how to classify different substances as one of the following: element, compound, homogeneous mixture, heterogeneous mixture, or suspension.



# Classification of Matter

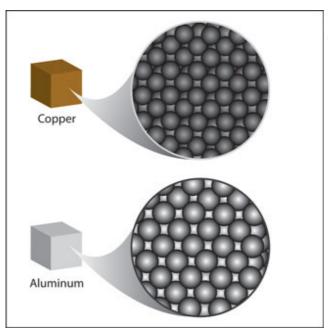
Element: refers to the substances on the periodic table. A sample of an element contains atoms which are all the same.

e.g. 100% pure copper wire



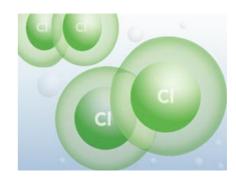
e.g. gold ingot





In a sample of an element all of the atoms are the same!

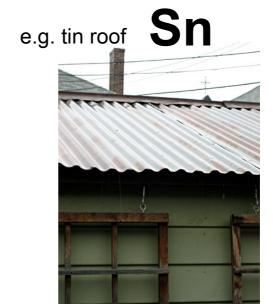
Cl<sub>2</sub> (gas)



e.g. iron nails

Fe





e.g. oxygen gas

 $O_2$ 



e.g. helium gas

He



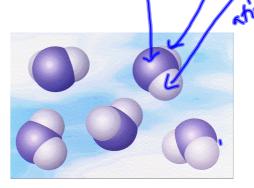
Compound: refers to a substance which contains molecules consisting of different kinds of atoms.

In a compound all of the molecules are

the same.

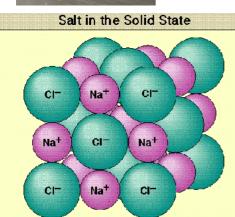






e.g. Salt, NaCl



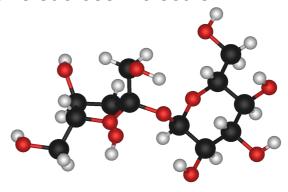


e.g. table sugar (sucrose),

 $C_{12}H_{22}O_{11}$ 



One sucrose molecule:



Mixtures: contain two or more different elements or compounds, just physically mixed together.

Two types of mixtures:

1 Homogenous mixtures

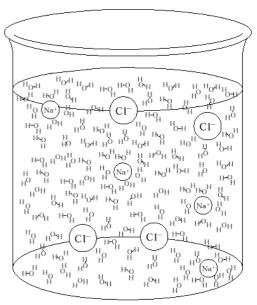
To the eye this mixtures looks like one substance. It looks uniform throughout.



e.g. seawater



e.g. salt water
It is a mixture of two compounds,
water and salt, but to the eye it
looks like it is just water. You can't
see the salt when it dissolves.
Salt water is completely clear,
just like pure water.

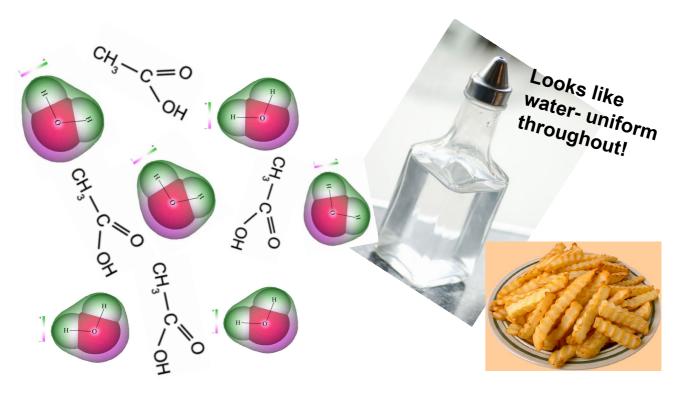


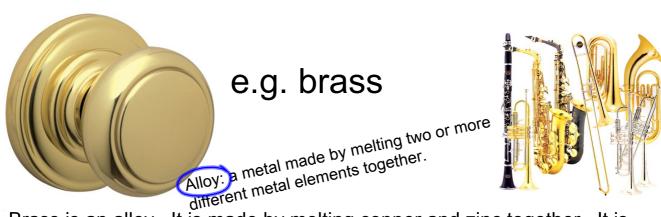
#### e.g. vinegar

vinegar is a mixture of acetic acid (CH₃COOH) and water (H₂O).

5% acetic acid, 95% water (by volume)







Brass is an alloy. It is made by melting copper and zinc together. It is used for decoration for its bright gold-like appearance; for applications where low friction is required such as locks, gears, bearings, doorknobs, ammunition, and valves; for plumbing and electrical applications; and extensively in musical instruments such as horns and bells for its acoustic properties. It is also used in zippers. Because it is softer than most other metals in general use, brass is often used in situations where it is important that sparks not be struck, as in fittings and tools around explosive gases.

Brass has a muted yellow color which is somewhat similar to gold. It is relatively resistant to rusting, and is often used as decoration and for coins. In antiquity, polished brass was often used as a mirror.

Atoms in an alloy more than one kind of atom mixed! :



## More examples of alloys:

Sterling silver: contains 92.5% silver and 7.5% copper This alloy has a bright surface!





Dental amalgam contains silver, tin, copper, and mercury

This alloy is easily worked!





→ Plumber's solder contains lead and tin

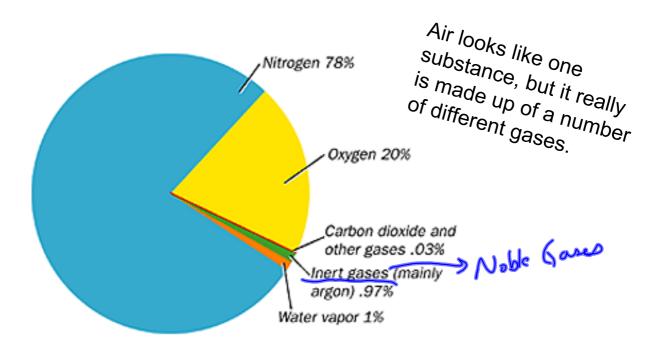
This alloy has a low melting point (275 deg C)

# **Stainless steel**

is an alloy of iron, carbon, chromium, and nickel
This alloy resists rusting!:



#### Another very interesting homogenous mixture: AIR !!!



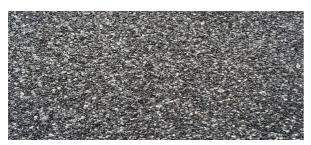
# 2. Heterogeneous mixtures

To the eye this mixture looks like two or more different substances.





#### More examples of heterogeneous mixtures:



asphalt (used to pave roads)

garden soil





concrete (contains cement gravel, and sand)





strawberry milkshake





fruitcake

# Last category: Suspensions

A suspension is a homogeneous liquid mixture.... with "flecks" added!

## The classic suspension examples:



OJ

tomato juice



Now let's practice! Draw a line to match each substance with its proper category term!

pure distilled













### And let's do it again!





Baking soda is sodium bicarbonate, NaHCO<sub>3</sub>

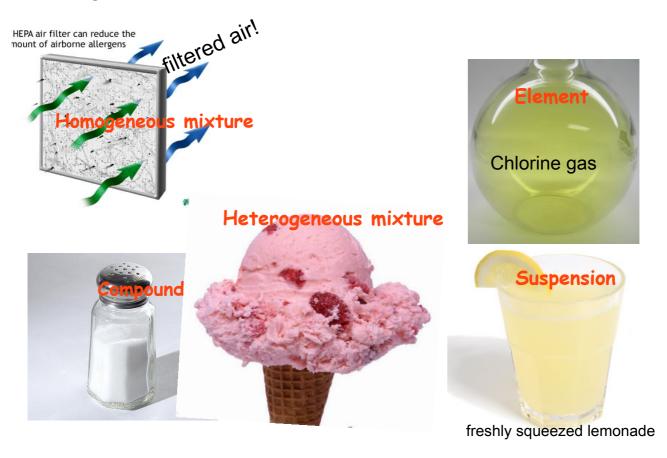


Largest polished diamond (solid carbon) in the world: The Golden Jubilee (109grams)





# And again!



## Now try to label these on your own:

	10 10 10 10 10 10 10 10 10 10 10 10 10 1
1.	garden soil Heteorgeneous Mixture
2.	filtered sea water   omogenous
3.	100% copper element
4.	tomato juice SUSPYNSION
5.	brass (alloy of copper and zinc) Homogeneous
6.	strawberry milkshake het mix
7.	24 carat gold bracelet (pure gold) (A) element
8.	sugar water homogeneous
9.	mineral water homogeneous
10	. ethyl alcohol (ethanol) (CHCH <sub>2</sub> OH)